THIA-ANALOGUES OF TROPANE ALKALOIDS. SYNTHESIS OF INTERMEDIATES FROM MONOCYCLIC PRECURSORS. Peter H. McCabe<sup>\*</sup> and William Routledge, Department of Chemistry, University of Glasgow, Glasgow, Gl2 8QQ. (Received in UK 24 Ostober 1975; accepted for publication 20 November 1975)

The trans addition of SC1<sub>2</sub> to double bonds has been applied to unsubstituted cyclic polyenes for the preparation of sulphur-bridged bicyclic systems.<sup>1,2</sup> SC1<sub>2</sub> also reacts with active methylene compounds with elimination of HCl to form acyclic thioethers (1:2 adduct)<sup>3</sup> and thioketones (1:1 adduct)<sup>4</sup>.

To achieve a general synthesis of 8-thiabicyclo[3.2.1]octanes, as precursors to 8-thiaanalogues of tropane alkaloids, we have investigated the condensation of SC1<sub>2</sub> with substituted cycloheptadienes,<sup>5</sup> two of which contain an enolisable carbonyl group.

Diene (1) reacted predictably with SC1<sub>2</sub> in CH<sub>2</sub>C1<sub>2</sub> at  $-70^{\circ}$  producing, in quantitative yield, dichlorobenzoate (2) [m.p. 160-161°; m/e 316, 318, 320 (M<sup>+</sup>)]. In contrast, dienone (3) behaved as an active methylene compound and reacted in CH<sub>2</sub>C1<sub>2</sub> at  $-5^{\circ}$  with evolution of HC1 giving an unstable pale yellow oil as major product (80%). This material, which possessed two distinct bridgehead resonances ( $\delta$  3.57 and 3.86) and a non-conjugated carbonyl group ( $\nu^{CC14}$ 1749 cm<sup>-1</sup>), was identified as allylic chloride (4) on the basis of double resonance experiments. During chromatography on alumina or on heating in vacuum 4 underwent partial decomposition ( $\leq$  10%) with further loss of HC1 forming the canary-yellow, photolabile enone (5) ( $\nu^{\text{KBr}}$ 1696 cm<sup>-1</sup>) by a skeletal rearrangement similar to that observed in 9-thiabicyclo[3.3.1]nonyl systems.<sup>6</sup>

In condensation with SCl<sub>2</sub> diene acetal (6) and dienone (7) offer more than one possible mode of addition. Surprisingly, each furnished one major adduct (> 90%) viz. § [m.p. 112-114°; m/e 254, 256, 258 (M<sup>+</sup>)] and 9 [m.p. 139-141°; m/e 210, 212, 214 (M<sup>+</sup>);  $v^{\text{CCl}4}$  1761 cm<sup>-1</sup>]. The structural assignment of 8 and 9 followed from their PMR spectra, each of which possessed two non-coincident <u>H-¢-Cl</u> resonances in the  $\delta$  4.5-5.0 region, inconsistent with more symmetrical systems such as <u>11</u> and <u>12</u>. Moreover, dilute aqueous acid selectively hydrolysed 8, with retention of the acetal function, to chloroalcohol (10) [m.p. 85-87°, m/e 236, 238 (M<sup>+</sup>);  $v^{\text{CHCl}3}$  3559 cm<sup>-1</sup>], reflecting the difference in reactivity of the two chlorine substituents of 8.

From the limited number of examples studied, the highly regio- and stereoselective addition of SC1<sub>2</sub> to substituted cycloheptadienes represents a convenient entry to the synthesis of the thio-ether relatives of the tropanes and furnishes intermediates with useful functionality. By comparison, a synthesis<sup>7</sup> of the 8-thiabicyclo[3.2.1] octane ring system using expensive 8-aza analogues as precursors proved to be unsuccessful for certain substrates.



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Satisfactory analytical data have been obtained for 2, 8, 9 and 10 (microanalysis) and 4 (mass measurement) but not for the unstable enone (5).

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## References

- (a) E.J. Corey and E. Block, <u>J. Org. Chem.</u>, <u>31</u>, 1663 (1966).
  (b) F. Lautenschlaeger, <u>ibid.</u>, <u>33</u>, 2627 (1968).
- 2. E.D. Weil, K.J. Smith and R.J. Gruber, *ibid.*, <u>31</u>, 1669 (1966).
- 3. A. Delisle, <u>Ber</u>., <u>20</u>, 2008 (1887).
- 4. K.G. Naik and S.A. Vaishnav, J. Ind. Chem. Soc., 13, 25 (1936).
- 5. E.W. Garbisch, J. Org. Chem., 30, 2109 (1965).
- 6. P.H. McCabe and C.M. Livingston, Tetrahedron Letters, 3029 (1973).
- 7. V. Horak, J. Zavada and A. Piskala, Chem. and Ind., 1113 (1958).